5/19/2021 18CV44

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# CMR Institute of Technology, Bangalore DEPARTMENT OF CIVIL ENGINEERING I - INTERNAL ASSESSMENT

Semester: 4-CBCS 2018

Subject: CONCRETE TECHNOLOGY (18CV44)

Faculty: Mr Ruchir

Date: 20 May 2021

Time: 01:00 PM - 02:40 PM

Max Marks: 50

|      | Answer any 5 question(s)   |       |     |     |       |
|------|--|-------|-----|-----|-------|
| Q.No |  | Marks | СО  | РО  | BT/CL |
| 1    | What are Bogue's compounds? Briefly explain their contribution towards gaining of strength of cement.      | 10    | CO1 | PO1 | L2    |
| 2    | List Different types of cement. Explain properties and application of any two types cement in detail.      | 10    | CO1 | PO1 | L2    |
| 3    | What is admixture? Explain any two Chemical admixtures in detail   | 10    | CO1 | PO1 | L2    |
| 4    | Write the chemical composition of cement. Explain with flow chart for dry process of manufacture of cement | 10    | CO1 | PO1 | L2    |
| 5    | Explain the importance of Source, Size, Shape, Texture, and Grading of aggregate.                          | 10    | CO1 | PO1 | L2    |
| 6    | What are mineral admixtures? Explain the importance and effect of Fly ash and GGBS on Concrete             | 10    | CO1 | PO1 | L2    |

| .N | Solution   |  |  |  | S      |  |
|----|--|--|--|--|--------|--|
|    | The following are termed as Bouges compound  |  |  |  |        |  |
|    | Compound   | Formula  | Shorthand<br>form  | % by<br>weight <sup>1</sup>  |        |  |
|    | Tricalcium aluminate   | Ca <sub>3</sub> Al <sub>2</sub> O <sub>6</sub>   | C <sub>3</sub> A   | 10   |        |  |
|    | Tetracalcium aluminoferrite  | Ca <sub>4</sub> Al <sub>2</sub> Fe <sub>2</sub> O <sub>10</sub>  | C <sub>4</sub> AF  | 8  |        |  |
|    | Belite or dicalcium silicate   | Ca <sub>2</sub> SiO <sub>5</sub>   | C <sub>2</sub> S   | 20   |        |  |
|    | Alite or tricalcium silicate   | Ca <sub>3</sub> SiO <sub>4</sub>   | C <sub>3</sub> S   | 55   |        |  |
|    | Sodium oxide   | Na <sub>2</sub> O  | N  | )  |        |  |
|    | Potassium oxide  | K <sub>2</sub> O   | K  | )Up to 2   |        |  |
|    | Gypsum   | CaSO <sub>4</sub> .2H <sub>2</sub> O   | CSH <sub>2</sub>   | 5  |        |  |
|    | Tricalcium aluminate, C3A:- It liberates a lot of heat du little strength contribution of C <sub>3</sub> A. Cement low in C <sub>3</sub> A i Tricalcium silicate, C <sub>3</sub> S:- This compound hydrates a  | . Gypsum slov<br>s sulphate res  | vs down the histant.   | nydration rate   |        |  |
|    | It liberates a lot of heat du little strength contribution of C <sub>3</sub> A. Cement low in C <sub>3</sub> A   | . Gypsum sloves sulphate research hardens rated all set and early slowly. It is larged the rom 3,000° F to bute much to set appropriately                            | vs down the histant.  pidly. It is largy strength gain gely responsible melting tempts 2,600° F). It is trength of the ly, manufactu | gely responsible in.  ble for strength perature of the thydrates e cement paste. | e<br>1 |  |
|    | It liberates a lot of heat du little strength contribution of C₃A. Cement low in C₃A is Tricalcium silicate, C₃S:- This compound hydrates a for portland cement's initian Dicalcium silicate, C₂S: C₂S hydrates and hardens a gain after one week. Ferrite, C₄AF: This is a fluxing agent whi raw materials in the kiln (for rapidly, but does not contribute by mixing these compound produce different types of environments.  | . Gypsum sloves sulphate research hardens radius all set and early slowly. It is larged the rom 3,000° F to bute much to suit described appropriately cement to suit | vs down the histant.  pidly. It is largy strength gain gely responsible melting tempts 2,600° F). It is trength of the ly, manufactu | gely responsible in.  ble for strength perature of the thydrates e cement paste. | e<br>1 |  |
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- 5. Low Heat Cement
- 6. Sulfates resisting cement
- 7. Blast Furnace Slag Cement
- 8. High Alumina Cement
- 9. White Cement
- 10. Colored cement
- 11. Air Entraining Cement
- 12. Expansive cement
- 13. Hydrographic cement

## [1] Cement-Rapid Hardening Cement (IS 8041–1990)

- This cement is similar to ordinary Portland cement.
- As the name indicates it develops strength rapidly and as such it may be more appropriate to call it as high early strength cement.
- Rapid hardening cement develops at the age of three days, the same strength as that is expected of ordinary Portland cement at seven days.
- The rapid rate of development of strength is attributed to the higher fineness of grinding (specific surface not less than 3250 sq. cm per gram) and higher C3S and lower C2S content.
- A higher fineness of cement particles expose greater surface area for action of water and also higher proportion of C3S results in quicker hydration.
- Rapid hardening cement gives out much greater heat of hydration during the early period. Therefore, rapid hardening cement should not be used in mass concrete construction.
- The use of rapid heading cement is recommended in the following situations:
- In pre-fabricated concrete construction.
- Where formwork is required to be removed early for re-use elsewhere,
- Road repair works,
- In cold weather concrete where the rapid rate of development of strength reduces
- the vulnerability of concrete to the frost damage.

## [2] Extra Rapid Hardening Cement

- Extra rapid hardening cement is obtained by intergrinding calcium chloride with rapid hardening Portland cement.
- The normal addition of calcium chloride should not exceed 2 percent by weight of the rapid hardening cement.
- It is necessary that the concrete made by using extra rapid hardening cement should be transported, placed and compacted and finished within about 20 minutes.
- It is also necessary that this cement should not be stored for more than a month.
- Extra rapid hardening cement accelerates the setting and hardening process.
- A large quantity of heat is evolved in a very short time after placing. The acceleration of setting, hardening and evolution of this

|    | <ul> <li>the ceme</li> <li>The gain strength of cement in the corrosion used,</li> <li>but in general such there</li> </ul>  | nt very suitable for of strength will dis of extra rapid harde hay be nearly the sa some evidence that a of reinforcement wheral, this effect do   | early period of hydraconcreting in cold wappear with age and ening cement or the owne.  there is small amount when extra rapid hardes are not appear to be pure extra rapid harden.  | reather, at 90 days the rdinary portland at of initial dening cement is rogressive and as   |    |
|----|--|--|--|---|----|
| 3  | <ul> <li>Cement</li> <li>Water</li> <li>Aggregates</li> <li>Which are used as an before or during mixing</li> <li>1. Plasticizers (Water Resolution)</li> <li>The organic substances, we workability, on termed as planged</li> <li>Plasticizers are</li> <li>In order to present the difficult to mixing used.</li> <li>Retarders</li> <li>A retarder is an admixing that concrete remains the retarder.</li> <li>Retarders are on setting processing p</li></ul> | deducing Agents) substances or combinate thich allow a reduction of give a higher work a sticizing admixtures. The also called high randoduce stronger concurrency, which makes the accessitating the unserties of concrete in the concrete has to be calcium Sulphate are | ations of organic and into an in water content for ability at the same water reducers. The rete, less water is addense of plasticizers or water the chemical processe for a longer time that the accelerating effect on hot weather concret be place in very difficults. | norganic the given er content, are ed (without ess workable and ater reducers are s of hydration so n concrete without of high temperature ing. It conditions and | 10 |
| 4. | Chemical Composition   | of cement  |  |   | 10 |
|    | Compound Calcium oxide (lime) Silicon dioxide  | Formula Ca0 SiO <sub>2</sub>   | Shorthand form C S   | Percent content<br>60–67<br>17–25   |    |
|    | (silica) Aluminum oxide (alumina)  | Al <sub>2</sub> O <sub>3</sub>   | A  | 3.0-8.0   |    |
|    | Iron oxide   | Fe <sub>2</sub> O <sub>3</sub>   | F  | 0.5–6.0   |    |

| Magnesium<br>Oxide                   | MgO                  | M   | 0.1–4.0 |
|--------------------------------------|----------------------|---|---------|
| Sulphate                             | $SO_3$               | S   | 1.3–3.0 |
| Alkalies                             | (K2O, Na2O)          |   | 0.4–1.3 |
| Calcareous<br>materials  Jaw Crusher |                      | Argillaceous<br>materials  Jaw Crusher  Bin |         |
|                                      | Mixing<br>Ilverizing |   |         |
| BIII                                 | Gypsum               | Rotary<br>Kiln                              | Air     |
| Portl                                |                      |   |         |
|                                      |                      |   |         |
| Manufacture of cemen                 | t by Dry process     |   |         |

## 5.

Most igneous rocks make highly satisfactory concrete aggregates because they are normally hard, tough and dense. The igneous rocks have massive structure, entirely crystalline

or wholly glassy or in combination in between, depending upon the rate at which they were

cooled during formation. They may be acidic or basic depending upon the percentage of silica content. They may occur light coloured or dark coloured. The igneous rocks as a class are the most chemically active concrete aggregate and show a tendency to react with the alkalies in cement

## **Aggregates from Sedimentary Rocks**

The quality of aggregates derived from sedimentary rocks will vary in quality depending upon the cementing material and the pressure under which these rocks are originally compacted. Some siliceous sand stones have proved to be good concrete aggregate. Similarly, the limestone also can yield good concrete aggregate

## **Aggregates from Metamorphic Rocks**

Both igneous rocks and sedimentary rocks may be subjected to high temperature and pressure which causes metamorphism which changes the structure and texture of rocks. Metamorphic rocks show foliated structure. Many metamorphic rocks particularly quartizite and gneiss have been used for production of good concrete aggregates.

## Size

The largest maximum size of aggregate practicable to handle under a given set of conditions should be used. Perhaps, 80 mm size is the maximum size that could be conveniently used for concrete making. Using the largest possible maximum size will result in (i) reduction of the cement content (ii) reduction in water requirement (iii) reduction of drying shrinkage

However, the maximum size of aggregate that can be used in any given condition may be limited by the following (i) Thickness of section; (ii) Spacing of reinforcement;

(iii ) Clear cover; (iv ) Mixing, handling and placing techniques conditions **Shape** 

The shape of aggregates is an important characteristic since it affects the workability of concrete. From the standpoint of economy in cement requirement for a given water/cement ratio, rounded aggregates are preferable to angular aggregates.

But the angular aggregates are superior to rounded aggregates from the following two points of view-1) Angular aggregates exhibit a better interlocking effect in concrete, which property

makes it superior in concrete used for roads and pavements. 2) The total surface area of rough textured angular aggregate is more than smooth rounded aggregate for the given volume. By having greater surface area, the angular aggregate may show higher bond strength than rounded aggregates.

## **Texture**

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- Surface texture is the property, the measure of which depends upon the relative degree to which particle surfaces are polished or dull, smooth or rough.
- Surface texture depends on hardness, grain size, pore structure, structure of the rock, and the degree to which forces acting on the particle surface have smoothed or roughend it.
- As surface smoothness increases, contact area decreases, hence a highly polished particle will have less bonding area with the matrix than a rough particle of the same volume.
- A smooth particle, however, will require a thinner layer of paste to lubricate its movements with respect to other aggregate particles. It will, therefore, permit denser packing for equal workability and hence, will require lower paste content than rough particles.
- It has been also shown by experiments that rough textured aggregate develops higher bond strength in tension than smooth textured aggregate.
- Mineral admixtures refer to the finely divided materials which are added to obtain specific engineering properties of cement mortar and concrete.

#### Fly ash

- It is obtained from coal fired power plants.
- Fly ash is generally captured by Electrostatic precipitators or other particle filtration equipment before the flue gases reach the chimneys from boilers.

#### It is also known as coal ash:

- There are two ways that the fly ash can be used:
  - one way is to inter grind certain percentage of fly ash with cement clinker at the factory to produce Portland pozzolana cement (PPC) and
  - the second way is to use the fly ash as an admixture at the time of making concrete at the site of work.

The latter method gives freedom and flexibility to the user regarding the percentage addition of fly ash

#### Effect on fresh concrete:

- The setting time is increased when fly ash is used.
- Workability and flow of concrete are increased due to the spherical shape of the fly ash particles, which lends a ball-bearing type effect on the concrete mixture.
- Bleeding and segregation are usually reduced for well-proportioned fly ash concrete.
- The paste volume is increased when mass replacement of cement by fly ash is done.

#### **GGBS**

- Blast furnace slag is a by-product of the extraction of iron from iron ore.
   Coke and limestone are added as fluxes inside the blast furnace.
- In India, we produce about 7.8 million tons of blast furnace slag.
- Chemical composition of GGBFS The amount of CaO(40%) in slag determines its cementitious properties.
- Silicon dioxide 30-38%
- Aluminium oxide 15-25%.
- Apart from delaying the initial set, slag does not significantly alter the fresh concrete properties.
- The workability of slag concrete is similar to an equivalent PC concrete, primarily because slag possesses the same level of fineness as PC.
- The rate of strength gain is slowed down considerably when cement is replaced by slag.
- The ultimate strengths with slag are generally improved;
- The durability is also improved with the replacement of cement by slag.