CMR INSTITUTE OF TECHNOLOGY

USN					



## Internal Assesment Test - I

Sub:	Material Science Code:						: 1	15ME32A				
Date:	18 / 09 / 2017	Duration:	90 mins	Max Marks:	50	Sem:	III	Branc	ch:	Mechanical		
Answer ALL FIVE Questions												
							Marks	OBE				
							wan	CO	RBT			
1.	1. Derive the atomic packing factor for a hexagonal close packed structure.								[10]	CO1	L3	
<ul> <li>To produce a p-type semiconductor, boron is doped in pure silicon. Doping is done by B<sub>2</sub>O<sub>2</sub> vapour. The atmosphere is equivalent to a surface concentration of 3X10<sup>26</sup> boron atoms per cubic meter. Calculate the time required to get a boron content of 10<sup>23</sup> atoms per cubic meter at a depth of 2.5μm. The doping temperature is 1100°C and D at this temperature is 4X10<sup>-17</sup>m<sup>2</sup>/s.</li> </ul>								[10]	CO1	L3		
3.	Define fatigue failure. With diagrams explain three types of fatigue loading.								[10]	CO2	L1	
	<ul> <li>a. With a figure show how you determine offset yield point on a stress-strain curve.</li> <li>b. Define unit cell, space lattice, atomic packing factor and co-ordination number with respect to crystal structure.</li> </ul>								[6]	CO2	L1	
								er with	[4]	CO1	L1	
	5. Derive an expression to show that maximum shear stress is experienced by a slip plane at an angle of 45° with respect to the force applied.								[10]	CO2	L3	

Material Science 15ME32A. IAT-1 solution (2017-18). Hexagonal closed packed structuret 9.1 No of atoms per unit cell! APF = Volume of spheres = Vs Volume of spheres = Vu V5= 4x R3x6 => 3x (2) 6 To calculate Vul Area of base = area of 6 equilateral D'es AYER= 6X DABO.

Area = 6x2 xABX00 = 6x1 x Aa x00 NOW COS30° = 00' = 00' =) 00'=a(0330° =) 00'= C153 =) Areo = 6x \frac{1}{2} \times \frac{3}{2} = \frac{3\frac{3}{3}}{2} a^2 But volume = area × height. 3) Valume = 353 a x 6. 3) A From DA'AB, COS30'S AB' AA = AB (0830° = a )3 AX= 3 AA ) > AX= 3 5 2 = 3 ACZ=AXZ+XCZ >) \are - (\frac{1}{2})^2 + (\frac{1}{2})^2 =) \are - \frac{2}{3} + \frac{2}{4} シニューション ニューションニューラ >> \langle = 1.633 a) Vu= 353 a2 x 1.633 a.

2 Ginent-  

$$C_s = 3 \times 10^{26} \text{ atoms}/m^3$$
  
 $C_x = 10^{23} \text{ atoms}/m^3$   
 $C_0 = 0 \text{ atoms}/m^3$   
 $C_1 = 100^{\circ}C = 1373K$ .  
 $C_1 = 100^{\circ}C = 1373K$ .  
 $C_1 = 100^{\circ}C = 100^{\circ}C$   
 $C_1 = 100^{\circ}C$   

exf (3) 0.9993 0.9996 2. 6 0.9998 By interpolation, 3-2.4 - 0.9996-0.9993 0.9998-0.9993 2.6-2.4 > 3 - 2.52. =) 197,64 = 2.52. =) St = 197.16 2.52 =) == 6151.04 second Q3. Fatigue failure is sound to occur in a material when it is subjected to cyclic loading and the load applied is less than its fracture load. Types of Patigue loading: @ Fully revirsed loading. Oregented loading. (i) Irregular loading,

@ Fully reversed loading! In this type of loading the stress level is taken to one extreme And of the spectrum, and then it is reversed completly to the other side of the spectrum to the same magnitude. Egs Rotating shafts. (i) Repeated loading time -In this type of loading the material is subjected to a max stress and then reduced to a minimum stress but of the some nature This can be either tensile of compressive in nature. The loads operate within this envelop.

Eg. Fuselage of an aircraft. 6 (ii) Ixrigular (saelingt time > This type of loading ocause when the matrial is usually left to face the element. like wind, in nature. Here the loading on the material is unpredictable. Eg!- Blades of windmill @ wings of an aircraft. - offset yield point. Strain must draw a line parallel to the linear portion

of the curve at an offseting distance of 2% of the entire stooin value, where ever the line meets the curre will be the b) O Unit cell= It is the smallest reproduction block of the crystal. It repeats itself over large atomic distances to produce the crystal structure, (i) space lattices It is the 3-D array of points, each, point representing an

atom Every atom in a particular space latice vill have the same surroundings as any other point in the some space

(i) APTI- It is the space occupied by atoms in pa anit cell. It is given by APF = Volume of outong Volume of unit cell.

(iv) Co-ordination number

It is the number of money adjacent adjacent actions to any given atom. It remains some for levery atom in that particular grace lattice.

